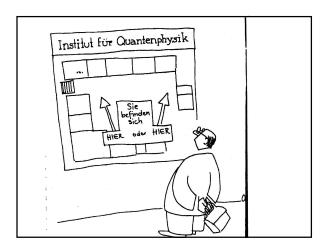
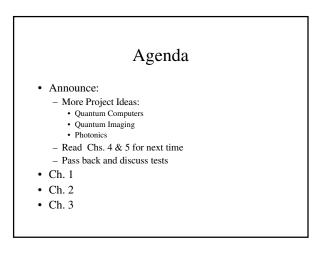


Ford: Chs 1-3







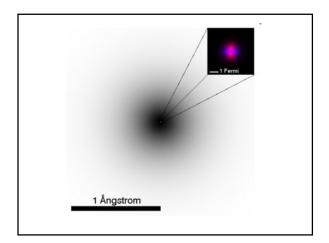


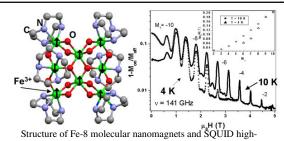
Ch. 1

- Atoms mostly empty space (e.g. propeller)
- Atoms very big compared to subatomic world
- Existence of atoms still in doubt in early 1900s
- Some particles: electron, proton, photon
- Standard Model: quarks, Higgs, leptons, force carriers
- · Classical determinism vs quantum probabilities

Ch. 2—Small/Fast

- Scientific Notation (of necessity)
- Range of units (eV, angstroms, etc)
- Physical Constants (h, hbar, c)
- What does it mean to be quantized?
- Charge:
 - Fundamental (property & force)
 - We're not sure why it's quantized
 - Holds electrons in atom
 - Balances strong force (gluons) inside nucleus
- Spin
 - A measure of angular momentum (different than orbital)
 - Fundamental property of particles (half-integer, and integer)
 - quantized

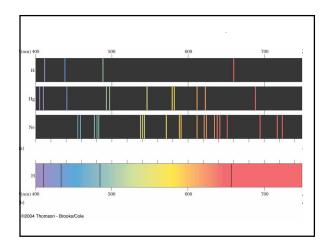


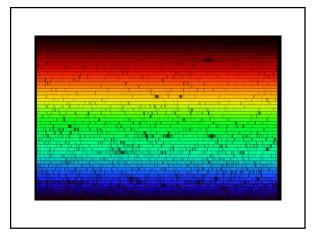


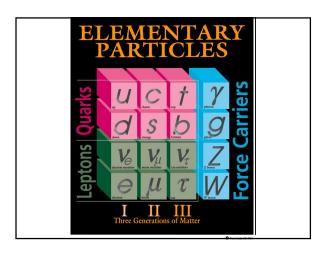
Structure of re-8 molecular nanomagnets and SQUID nighfrequency EPR spectra of Fe-8 showing resonant absorption corresponding to transitions between the quantized energy levels.

From:

http://www.boulder.nist.gov/div818/81803/2005/MagneticThinFilms/index.html







Ch. 3—The Leptons Antiparticles/antimatter—particles identical to their partners, aveant the approximate charge (same)

- their partners except the opposite charge (same mass and spin)
- Electron (1897 JJ Thompson)
 - First fundamental particle discovered
 - Basis of all electronics (but for how much longer?)

• Leptons

Neutrinos

- 3 flavors
- Only weakly interacting—no charge
- Hard to detect
- Very little masses



Muon

- 200x more massive than electron
- Otherwise very similar to electron
- Decays

Tau

- Very massive (more than proton)
- Same charge as muon and electron